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# Note Taking Guide

## Episode 702

### Answers

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Answer Key 702 Note Taking Guide - Kora

Example Problem: Find the % by mass of oxygen in water.

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**Curious Beginnings | Critical Role: THE MIGHTY**  
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guide episode 702 ...**

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When I Read Books

Book Margin Visual

Note-Taking Solar

Photovoltaic (PV)

Systems, Scope

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**the Most Out of Your**

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## **Taking Notes: Crash Course Study Skills**

**#1 Curious Beginnings | Critical Role: THE MIGHTY NEIN | Episode 1** **Note Taking Guide**

Episode 702 **Note Taking Guide: Episode 702 Name \_\_\_\_\_**

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Example Problem: Find  
the % by mass of  
oxygen in water.  
Percentage by mass of  
element in a compound  
 $= (\text{mass of element in } 1 \text{ mol of compound} \div \text{molar mass of compound}) \times 100 \%$   
(after completing lab  
#1) Find the % of  
carbon in sodium  
bicarbonate ( $\text{NaHCO}_3$ ).  
Find the %  
composition of  
aluminum oxide.Note  
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 Find the % by mass of  
 oxygen in water.  
 Percentage by mass of  
 element in a compound  
 = (mass of element in  
 1 mol of compound ÷  
 molar mass of  
 compound ) x 100 %

(after completing lab  
 #1)Example Problem:  
 Find the % by mass of  
 oxygen in water.Note  
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| Critical Role: THE  
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### **Note Taking Guide**

#### **Episode 702**

#### **Answers**

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\_\_\_\_\_ Example

Problem: Find the % by  
mass of oxygen in

water. Percentage by  
mass of element in a

compound = (mass of  
element in 1 mol of  
compound ÷ molar

mass of compound ) x  
100 % (after

completing lab #1)

Find the % of carbon in  
sodium bicarbonate

(NaHCO<sub>3</sub>). Find the %  
composition of

aluminum oxide.

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Guide Episode 1001*

### *Answers*

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 $88.9\%$  Percenta g e by  
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 mass of compound ) x  
 100 % (after  
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 Percentage by mass of  
 element in a compound  
 = (mass of element in  
 1 mol of compound ÷  
 molar mass of  
 compound ) x 100 %  
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**Example Problem:**  
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**of oxygen in water.**

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